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# Introduction

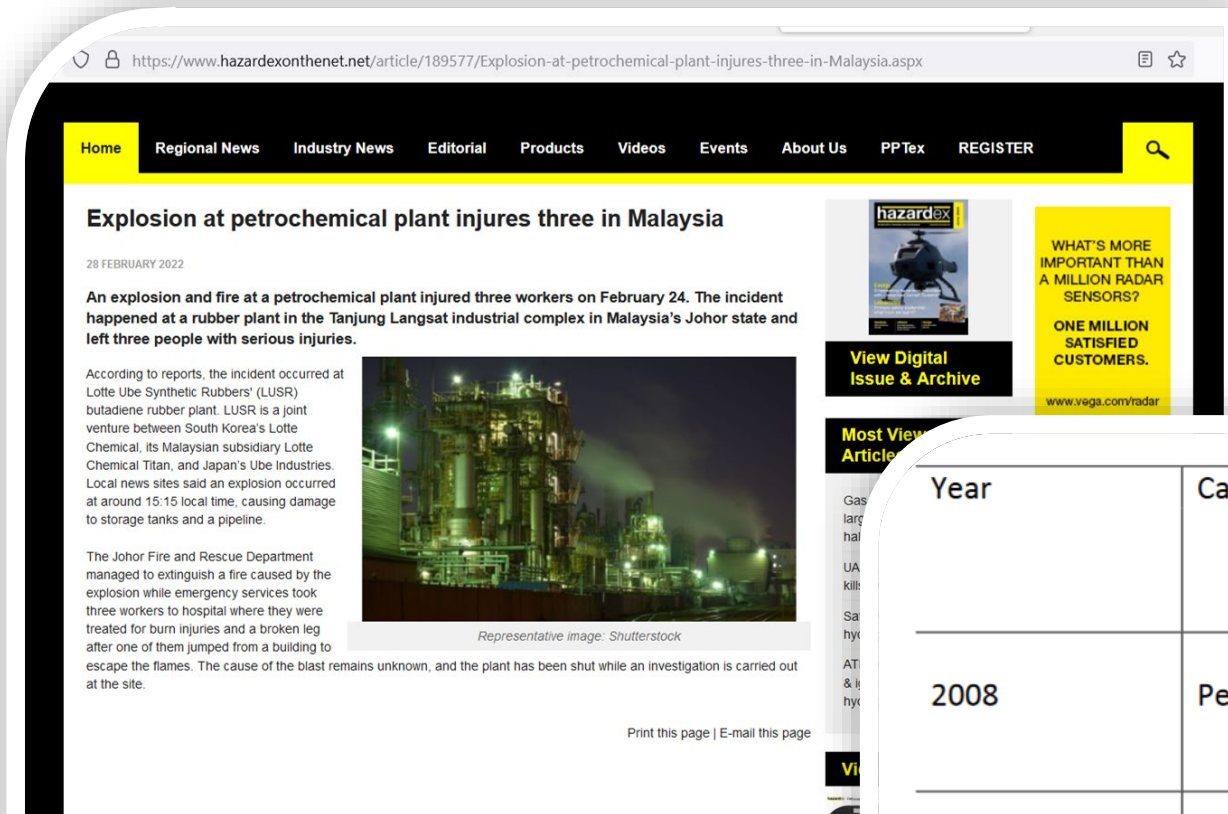
## Fire and Explosion in Chemical Industries

SIMT-ITS, 12 Juni 20234

**Juwari, S.T., M.Eng PhD**



# Several fire and or explosion incidents in Malaysia



Year	Case	Process Safety management (PSM) Elements			
		Process hazard Analysis	Training of Manpower	Emergency Preparedness & Response	
2008	Perak	✓	✓	✓	
2010	Pulau Pinang	✓	✓	✓	
2013	Pulau Pinang	✓	✓	✓	
2014	Pahang	✓	✓		

Table 1: PSM analysis for Malaysia Dust Explosion Incident





# Lesson learn CSB.GOV

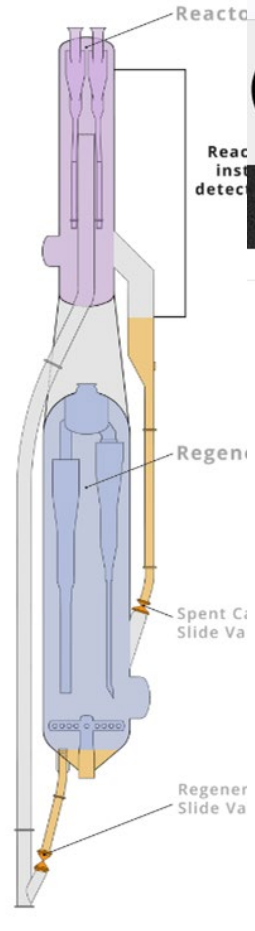


Figure 37. Potential undetectable catalyst levels within regenerator during a shutdown (person for scale). (Credit: CSB)

Screenshot of the CSB.GOV Videos page. The page features the CSB logo, navigation links (About the CSB, Investigations, Recommendations, Advocacy Priorities, Media Room), and a 'Videos' section. A video titled 'Transient Hazards: Explosion at the Husky Superior Refinery' is highlighted, with a description of the April 2018 explosion and fire. The page also includes a 'Statistics' section with a bar chart showing the number of videos in different categories.

**Statistics**

Category	Count
Uncovered Hazards	240
Animation of April 26, 2021	125
Blowout in Oklahoma	704

**Transient Hazards: Explosion at the Husky Superior Refinery**  
Friday, Jun 09 2023  
A CSB safety video about the April 2018 explosion and fire at the Husky Superior Refinery in Superior, Wisconsin. The incident injured 36 workers, caused roughly \$550 million in damage to the facility, and released 39,000 pounds of flammable hydrocarbon vapor into the air.

**Investigations:**

- Husky Energy Superior Refinery Explosion and Fire

**Video Search**

Browse By: Categories

You are viewing: All videos

**Video Grid:**

- Transient Hazards: Explosion at the Husky Superior Refinery
- Animation of 2018 Ethylene Release and Fire at Kuraray America in
- Ignored Warnings: Explosion in St. Louis
- Wake Up Call: Refinery Disaster in Philadelphia
- Silent Killer: Hydrogen Sulfide Release in Odessa, Texas
- CSB PUBLIC BUSINESS MEETING April 28, 2022 2pm EDT
- Incompatible Chemicals: Explosion at AB Specialty Silicones
- Simultaneous Tragedy: Fire at Evergreen Packaging
- Quarterly Business Meeting - January 26, 2022
- Quarterly Business Meeting - October 29, 2021
- CSB July 29th Safety Air Meeting
- Message from the Chairperson
- Aghorn Public Board Meeting
- Updated BP Texas City Animation on the 15th Anniversary of the
- Uncovered Hazards
- Animation of April 26,
- Blowout in Oklahoma
- Preliminary Animation of
- Emergency Response
- Winterization Safety
- Interim Animation of

<https://youtu.be/sFhKzK7jkKg>



# Fire Triangle



When fuel, oxidizer, and an ignition source are present at the necessary levels, burning will occur



A fire will not occur if

Fuel is not present or is not present in sufficient quantities

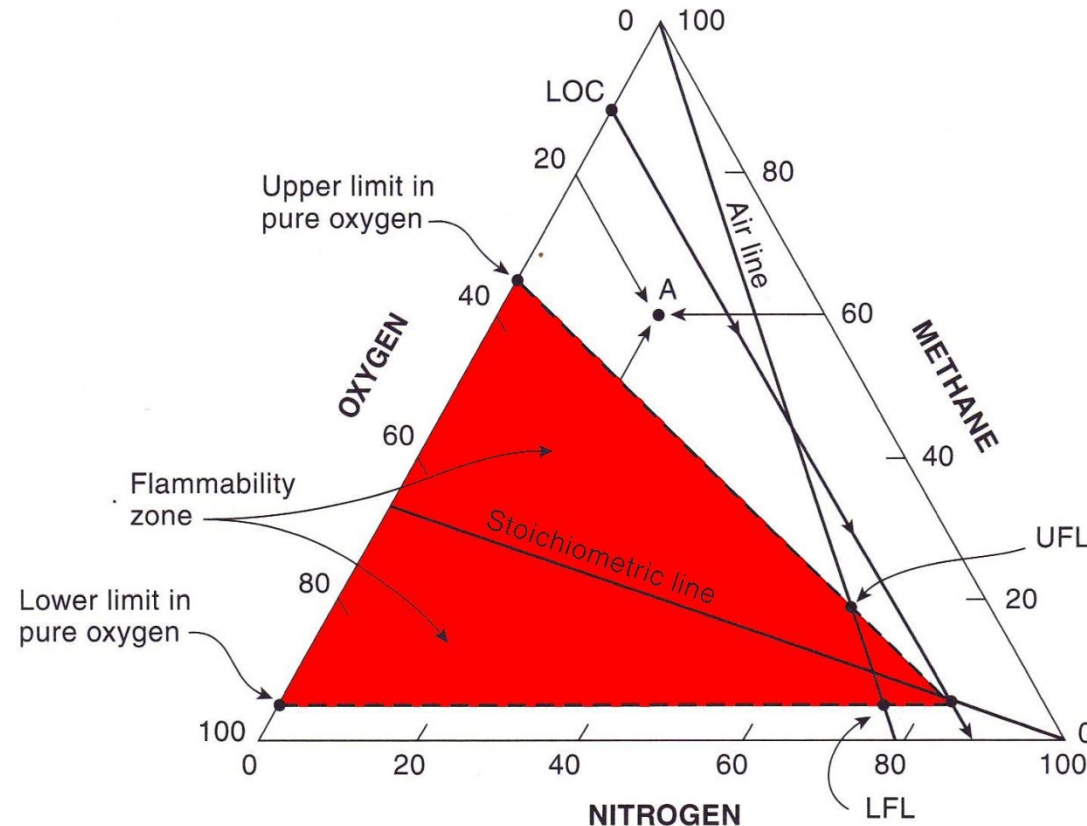
An oxidizer is not present or is not present in sufficient quantities

the ignition source is not energetic enough to initiate the fire



# Flammability Diagram

**Flammability diagrams** show the regimes of flammability in mixtures of fuel, oxygen and an inert gas, typically nitrogen.



- Useful for
  - Determining if a mixture is flammable
  - Required for control and prevention of flammable mixtures
  - Red zone – flammable zone
- The air line represent all possible combination of fuel plus air.
- The stoichiometric line represent all stoichiometric combination of fuel plus oxygen
- The LOC any gas mixture containing oxygen below the LOC is not flammable.

**Stoichiometric** is the calculation of quantities relationship of the reactants and products in a balanced chemical reaction. It can be used to calculate quantities such as the amount of products that can be produced with given reactants and percent yield.



## Flammability limits for mixtures

- The flash point temperature is one of the major quantities used to characterize the fire and explosion hazard of liquids. (The flash point of liquid is the lowest temperature at which it gives off enough vapor to form an ignitable mixture with air).
- Vapor mixtures  
frequently Lower Flammability Limit (LFL) and Upper Flammability Limit (UFL) for mixture are needed. These mixture limits are computed using the following formulae;

### Flammability limits

$$LFL_{mix} = \frac{1}{\sum_{i=1}^n \frac{y_i}{LFL_i}}$$

$$UFL_{mix} = \frac{1}{\sum_{i=1}^n \frac{y_i}{UFL_i}}$$



# Ignition sources

## Known ignition sources for vapour mixes

- Electric sparks and arcs (from electrical circuits, motors, switches etc.);
- Mechanical sparks (from friction and falling objects);
- Static electrical sparks;
- Lightning;
- Flame (including flaring, boilers, smoking);
- Hot surfaces (including hot work, hot processing equipment, electrical equipment);
- Heat of compression;
- Chemical reactions (eg auto-ignition of oil-soaked lagging on hot piping); and
- High energy radiation, microwaves, etc.



- Strategies
  - Inerting
    - To reduce the oxygen or fuel concentration to below a target value using an inert gas
    - Nitrogen is the most common
  - Use flammability diagram



Fire fighting system filled with inert gas





- Strategies
  - Controlling static electricity
    - How did it happen?

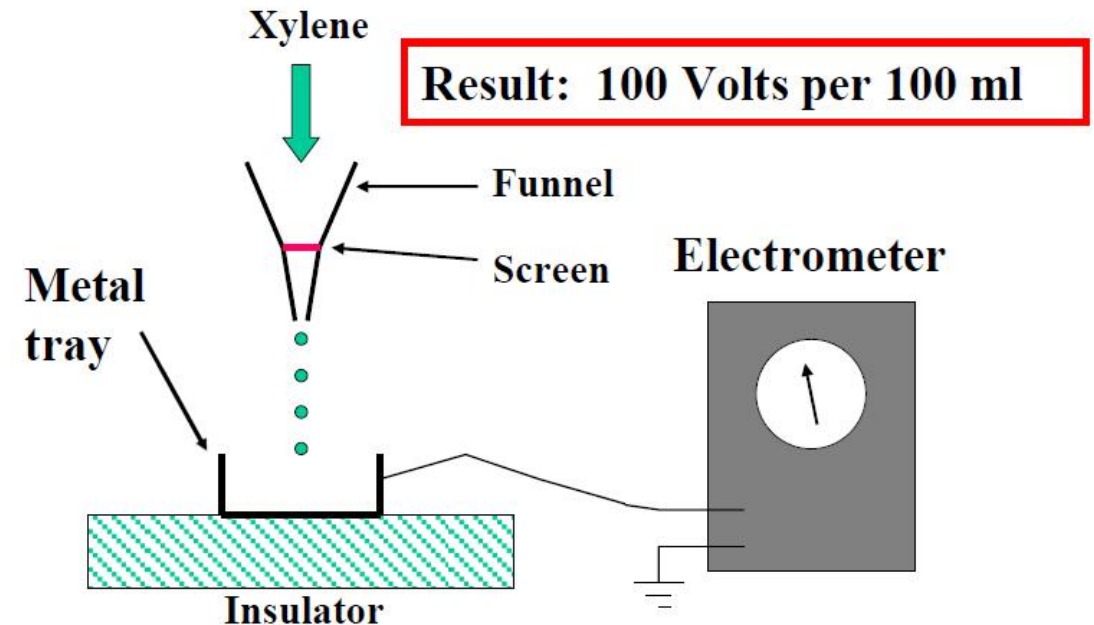


- By separation of two materials
- As materials are separated, electrons must redistribute themselves
- If materials are both conductors, electrons move rapidly
- If one or more materials are non-conductors, the electrons cannot move very fast and final result is a difference in charge
- Examples: **Rubbing, Falling, Moving, Flowing**



# Prevention

- Strategies
  - Controlling static electricity
    - Grounding and Bonding
    - Be careful of
      - Glass containers /vessels/pipes
      - Plastic containers /vessels/pipes/pumps
      - Low conductive liquids: benzene, toluene, xylene, heptane, hexane
    - Avoid free fall of liquids into vessels





# Prevention

- Strategies

- Ventilation

- For inside storage areas, use  $1 \text{ ft}^3/\text{min}/\text{ft}^2$  of floor area
    - For inside process areas, use  $1 \text{ ft}^3/\text{min}/\text{ft}^2$  or more of floor area

- Sprinkler system

- Closed Head: Typically found in occupied buildings
    - Open Head: Activated from a central location
    - Monitor nozzles: Fixed location, but can be directed
    - Water requirements:  $0.25 - 0.5 \text{ gpm}/\text{ft}^2$  protected

- Fire extinguisher

- Foam



# Prevention

ventilation



sprinkle



Fire extinguisher



foam







# Implementation of PSM (Process Safety Management) to make our plant safer

- The **proactive** and **systematic** identification, evaluation, and mitigation or prevention of chemical releases that could occur as a result of failures in process, procedures, or equipment.





# PSM Process Safety Management should be implemented well

- ❖ Process safety information
- ❖ Standard operating procedure
- ❖ Process hazard analysis
- ❖ Mechanical integrity



# Process Safety Information



**Employer must complete a compilation of written process safety**



**Identify & understand the hazards**



**PSI must include;**

- Info on hazards of chemicals used or produced
- Technology of the process
- Equipments used in the process
- Info on hazards of chemicals;
  - Toxicity
  - Permissible exposure limit
  - Physical data
  - Reactivity data
  - Corrosivity data
  - Thermal & chemical stability data
  - Effect of mixing with other chemicals



# Process Safety Information



## Info on technology of the process;

- Block flow diagram or simplified PFD
- Process chemistry
- Maximum intended inventory
- Safe upper & lower limits of T, P, x
- Evaluation of consequences of deviation



## Info on the equipments;

- Material of construction
- P&IDs
- Electrical classification
- Relief system design & design basis
- Ventilation system design
- Design codes & standard employed
- Safety system (interlocks, detection or suppression system)





## Sources and Process safety information

Two common examples of the three components of the fire triangle are

- wood, air, and a match
- gasoline, air, and a spark

Various fuels, oxidizers, and ignition sources common in the chemical industry are

- Fuels
  - Liquids: gasoline, acetone, ether, pentane
  - Solids: plastics, wood dust fibers, metal particles
  - Gases: acetylene, propane, carbon monoxide, hydrogen
- Oxidizers
  - Gases: oxygen, fluorine, chlorine
  - Liquids: hydrogen peroxide, nitric acid, perchloric acid
  - Solids: metal peroxides, ammonium nitrite
- Ignition sources
  - Sparks, flames, static electricity, heat



# Standard operating procedure

Employer must develop & implement written procedures ~ PSI

Appropriate, clear, consistent & well-communicated

SOP cover:-

- Initial startup and startup after (normal or emergency) shutdown – hourly.
- Normal and emergency operation
- Normal and emergency shutdown
- Operating limits and consequences of deviations
- SHE consideration – safe work practices, safety control systems and functions, hazards
- Quality control specification for all chemicals (raw material & product)
- Safety systems (interlocks, detection or suppression systems)
- Troubleshooting procedure.



# Standard operating procedure



**Must be ready & up-to-date for reference**



**Safe work practice for control of hazards during activities;**

Lockout/tagout

Confined space entry

Opening process equipment or piping

Control over entrance into a facility



# Process Hazard Analysis

## Examples of PHA methods;

- Hazard & operability study (HAZOP) – for more complex processes
- What-if scenarios
- Checklist
- What-if/Checklist
- Failure mode & effect analysis (FMEA)
- Fault tree analysis

## PHA must address the following;

- Hazard of the process
- Identification of previous incidents
- Engineering & administrative control & consequences of its failure
- Facility siting
- Human factors
- Qualitative evaluation of possible safety & health effects on employees





# Mechanical Integrity



**To ensure that the equipment, piping, relief systems, controls and alarms are mechanically sound and operational.**



**Applicable equipments;**

Pressure vessels & storage tanks  
Piping system (including valves)  
Relief & vent. systems & devices  
Emergency shutdown systems  
Controls (alarms, locks etc)  
Pumps



**Inspection & testing must be performed**



**Record must be kept;**

Date  
Name of person responsible  
Identifier of equipment (eg – serial no)  
Description of test  
Results of test



**Frequency – recommendation or more based on prior experience**



# Make Sure all elements in PSM implemented







**- TERIMA KASIH -**



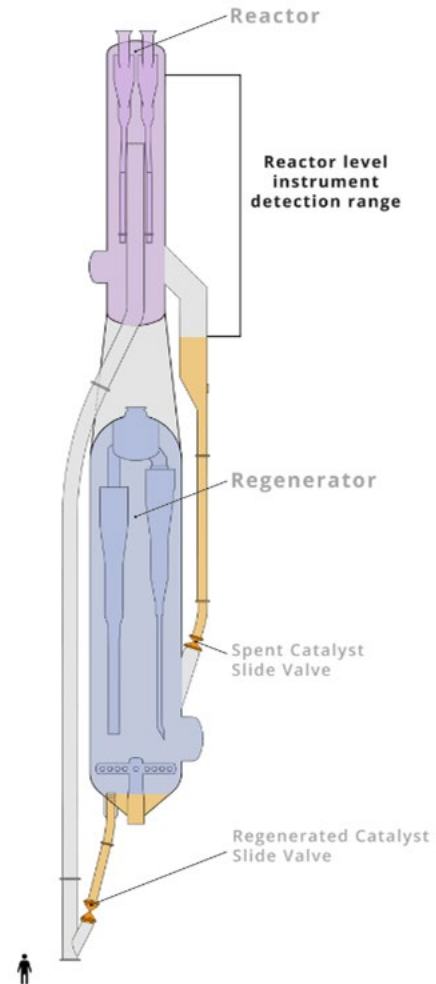
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**Figure 37.** Potential undetectable catalyst levels within the reactor and regenerator during a shutdown (person for scale). (Credit: CSB)